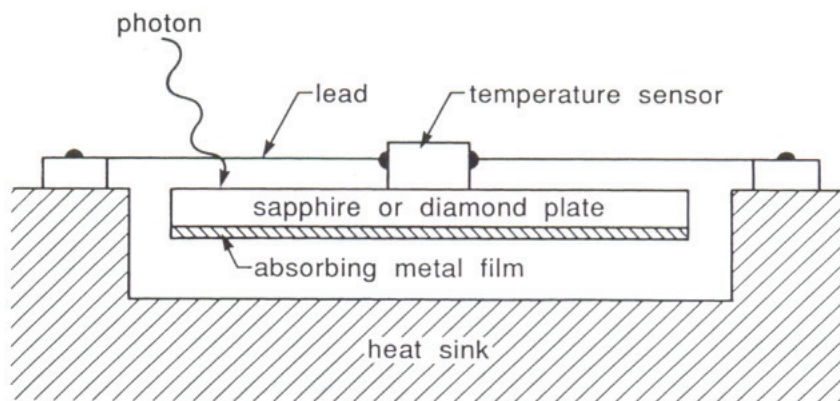


Astronomische Waarneemtechnieken (Astronomical Observing Techniques)

9th Lecture: 29 October 2012



Overview

I. PHOTOCONDUCTORS

Intro: Solid State Physics
Intrinsic Photoconductors
Extrinsic Photoconductors
Readout & Operations
Detector Noise
Flatfielding Techniques

II. BOLOMETERS

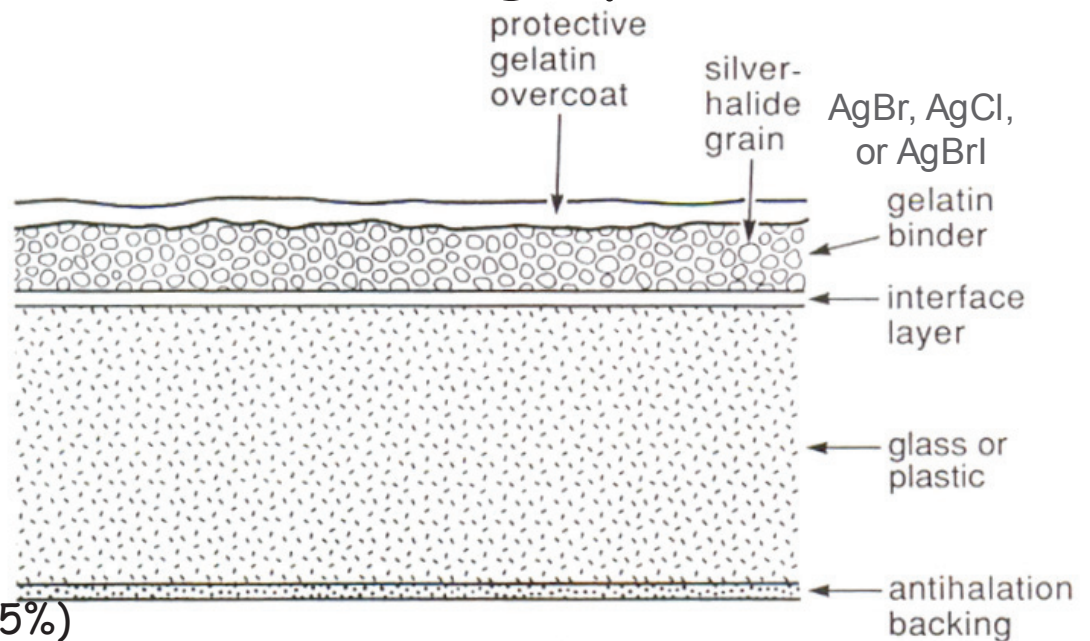
III. COHERENT RECEIVERS

Outlook: MKIDs

Based on "Detection of Light - from the Ultraviolet to the Submillimeter", by George Rieke, 2nd Edition, 2003, Cambridge University Press, ISBN 0-521-01710-6.

Overview

Preface: The Photographic Plate



Disadvantages:

- Low DQE (~2-5%)
- Non-linearity
- Non-uniformity
- Time resolution
- Wavelength coverage
- Digitization → image proc.

Advantages:

- 10" plate → 10^{12} grains → $\sim 10^9$ pixels
- Inexpensive
- Include own data storage system
- Stable over very long periods of time

Three Basic Types of Detectors

1. Photon detectors

Respond directly to individual photons → releases bound charge carriers. Used from X-ray to infrared.

Examples: photoconductors, photodiodes, photoemissive detectors

2. Thermal detectors

Absorb photons and thermalize their energy → modulates electrical current. Used mainly in IR and sub-mm detectors.

Examples: bolometers

3. Coherent receivers

Respond to electrical field strength and preserve phase information (but need a reference phase "local oscillator").
Mainly used in the sub-mm and radio regime.

Examples: heterodyne receivers

Part I

Photon Detectors

Part II

Thermal Detectors

Part III

Coherent Receivers

**Intro to
Solid State
Physics**

PERIODIC TABLE OF THE ELEMENTS

<http://www.ktf-split.hr/periodni/en/>

PERIOD	GROUP I IA	GROUP IIA	GROUP IIIA	GROUP IVB	GROUP VB	GROUP VIB	GROUP VIIB	GROUP VIIIB	GROUP IIB	GROUP IIIA	GROUP IVA	GROUP VA	GROUP VIA	GROUP VIIA	GROUP VIIIA			
1	1 1.0079 H HYDROGEN														2 4.0026 He HELIUM			
2	3 6.941 Li LITHIUM	4 9.0122 Be BERYLLIUM								5 10.811 B BORON	6 12.011 C CARBON	7 14.007 N NITROGEN	8 15.999 O OXYGEN	9 18.998 F FLUORINE	10 20.180 Ne NEON			
3	11 22.990 Na SODIUM	12 24.305 Mg MAGNESIUM								13 26.982 Al ALUMINIUM	14 28.086 Si SILICON	15 30.974 P PHOSPHORUS	16 32.065 S SULPHUR	17 35.453 Cl CHLORINE	18 39.948 Ar ARGON			
4	19 39.098 K POTASSIUM	20 40.078 Ca CALCIUM	21 44.956 Sc SCANDIUM	22 47.867 Ti TITANIUM	23 50.942 V VANADIUM	24 51.996 Cr CHROMIUM	25 54.938 Mn MANGANESE	26 55.845 Fe IRON	27 58.933 Co COBALT	28 58.693 Ni NICKEL	29 63.546 Cu COPPER	30 65.39 Zn ZINC	31 69.723 Ga GALLIUM	32 72.64 Ge GERMANIUM	33 74.922 As ARSENIC	34 78.96 Se SELENIUM	35 79.904 Br BROMINE	36 83.80 Kr KRYPTON
5	37 85.468 Rb RUBIDIUM	38 87.62 Sr STRONTIUM	39 88.906 Y YTRIUM	40 91.224 Zr ZIRCONIUM	41 92.906 Nb NIObIUM	42 95.94 Mo MOLYBDENUM	43 (98) Tc TECHNETIUM	44 101.07 Ru RUTHENIUM	45 102.91 Rh RHODIUM	46 106.42 Pd PALLADIUM	47 107.87 Ag SILVER	48 112.41 Cd CADMIUM	49 114.82 In INDIUM	50 118.71 Sn TIN	51 121.76 Sb ANTIMONY	52 127.60 Te TELLURIUM	53 126.90 I IODINE	54 131.29 Xe XENON
6	55 132.91 Cs CAESIUM	56 137.33 Ba BARIUM	57-71 La-Lu Lanthanide	72 178.49 Hf HAFNIUM	73 180.95 Ta TANTALUM	74 183.84 W TUNGSTEN	75 186.21 Re RHENIUM	76 190.23 Os OSMIUM	77 192.22 Ir IRIDIUM	78 195.08 Pt PLATINUM	79 196.97 Au GOLD	80 200.59 Hg MERCURY	81 204.38 Tl THALLIUM	82 207.2 Pb LEAD	83 208.98 Bi BISMUTH	84 (209) Po POLONIUM	85 (210) At ASTATINE	86 (222) Rn RADON
7	87 (223) Fr FRANCIUM	88 (226) Ra RADIUM	89-103 Ac-Lr Actinide	104 (261) Rf RUTHERFORDIUM	105 (262) Db DUBNIUM	106 (266) Sg SEABORGIUM	107 (264) Bh BOHRNIUM	108 (277) Hs HASSIUM	109 (268) Mt MEITNERIUM	110 (281) Uun UNUNNIUM	111 (272) Uuu UNUNUNIUM	112 (285) Uub UNUNBIUM						

RELATIVE ATOMIC MASS (A)

GROUP IUPAC

GROUP CAS

ATOMIC NUMBER

SYMBOL

ELEMENT NAME

Legend:

- Metal
- Semimetal
- Nonmetal
- Alkali metal
- Alkaline earth metal
- Transition metals
- Lanthanide
- Actinide
- Chalcogens element
- Halogens element
- Noble gas

STANDARD STATE (25 °C; 101 kPa)

Ne - gas Fe - solid
Ga - liquid Tc - synthetic

LANTHANIDE

57 138.91 La LANTHANUM	58 140.12 Ce CERIUM	59 140.91 Pr PRASEODYMIUM	60 144.24 Nd NEODYMIUM	61 (145) Pm PROMETHIUM	62 150.36 Sm SAMARIUM	63 151.96 Eu EUROPIUM	64 157.25 Gd GADOLINIUM	65 158.93 Tb TERBIUM	66 162.50 Dy DYSPROSIUM	67 164.93 Ho HOLMIUM	68 167.26 Er ERBIUM	69 168.93 Tm THULIUM	70 173.04 Yb YTTERIUM	71 174.97 Lu LUTETIUM
------------------------------	---------------------------	---------------------------------	------------------------------	------------------------------	-----------------------------	-----------------------------	-------------------------------	----------------------------	-------------------------------	----------------------------	---------------------------	----------------------------	-----------------------------	-----------------------------

ACTINIDE

89 (227) Ac ACTINIUM	90 232.04 Th THORIUM	91 231.04 Pa PROTACTINIUM	92 238.03 U URANIUM	93 (237) Np NEPTUNIUM	94 (244) Pu PLUTONIUM	95 (243) Am AMERICIUM	96 (247) Cm CURIUM	97 (247) Bk BERKELIUM	98 (251) Cf CALIFORNIUM	99 (252) Es EINSTEINIUM	100 (257) Fm FERMIUM	101 (258) Md MENDELEVIUM	102 (259) No NOBELIUM	103 (262) Lr LAWRENCIUM
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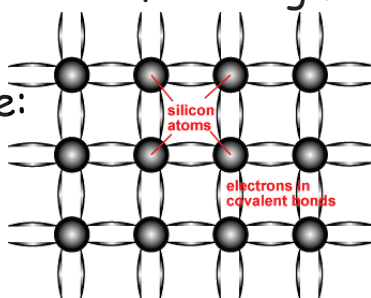
(1) Pure Appl. Chem., 73, No. 4, 667-683 (2001)
Relative atomic mass is shown with five significant figures. For elements having no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element.
However three such elements (Th, Pa, and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

Editor: Aditya Vardhan (advivar@netlinx.com)

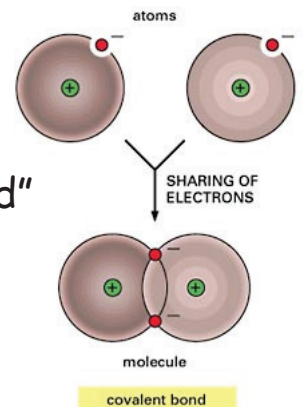
The Diamond Lattice

Elements with 4 e^- form crystals with **diamond lattice structure** (each atom bonds to four neighbors).

Example:

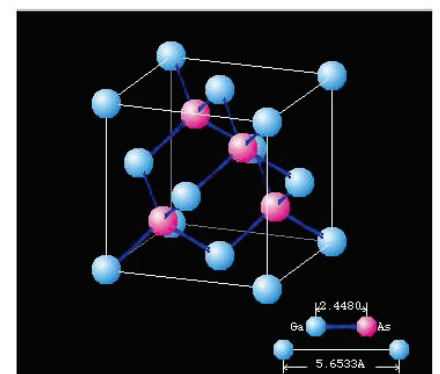
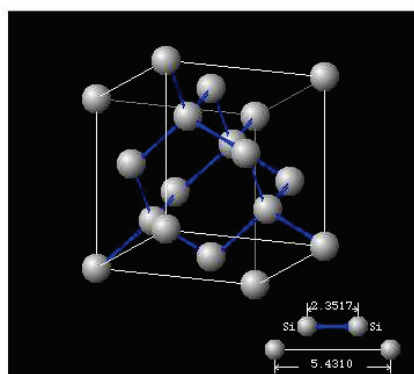


These double-bonds between neighbours are due to "shared" electrons



Note that a diamond lattice can not only be formed by IV elements [C, Si, Ge] but also by III-V semiconductors

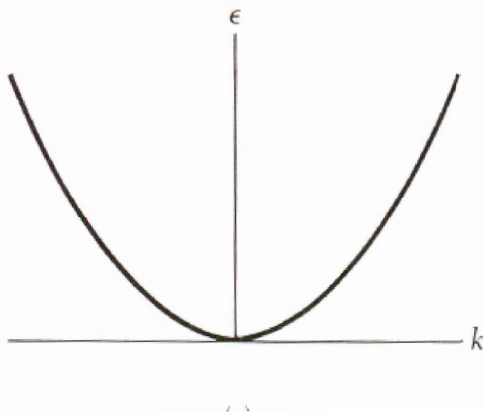
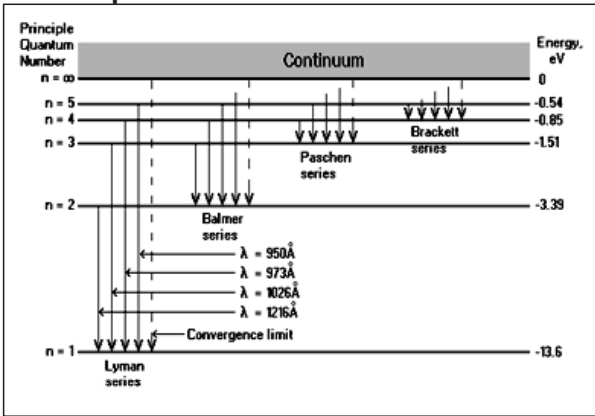
	IIIA	IVA	VA	VIA
	5 B	6 C	7 N	8 O
	13 Al	14 Si	15 P	16 S
	30 Zn	32 Ga	33 Ge	34 As
	48 Cd	49 In	50 Sn	52 Sb



Electronic States and Bands

Single atomic system

Example: H atom

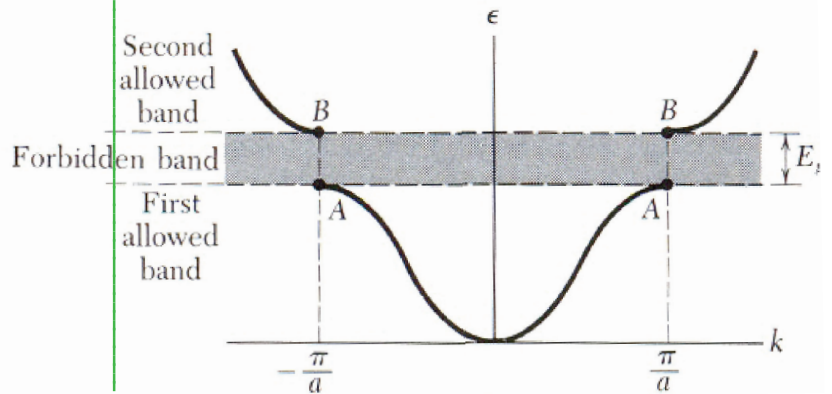


Atomic crystal

Wavefunctions Ψ overlap

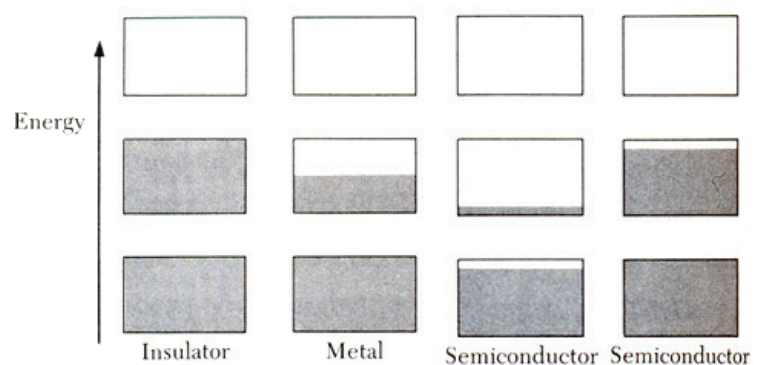
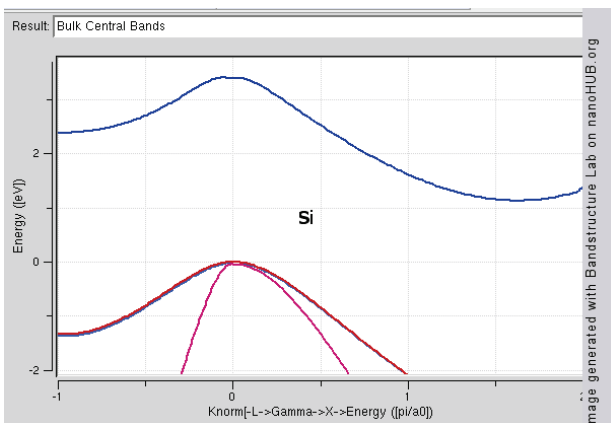
→ Energy levels of individual atoms split due to Pauli principle (avoiding the same quantum states)

→ Multiple splitting → "bands"



Electric Conductivity

Conductivity requires charge carriers in the conduction band

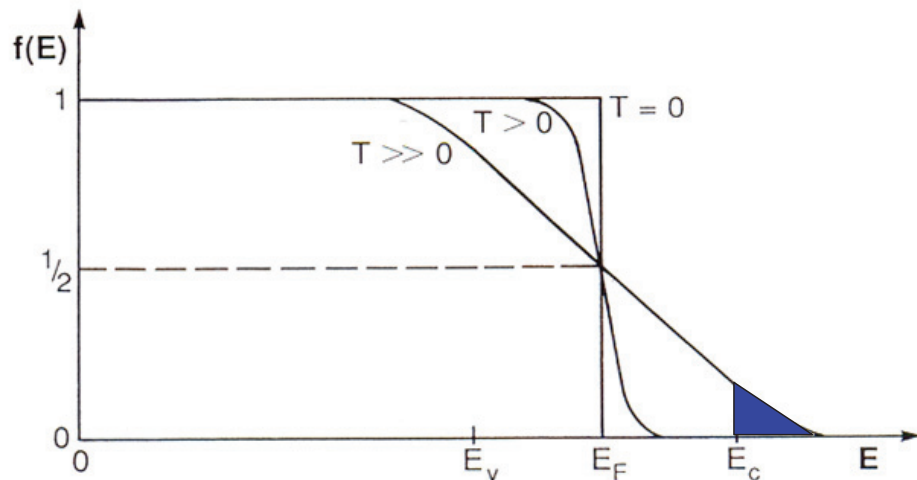


One needs to overcome the bandgap E_g to lift an e^- into the conduction band. This can be done via:

1. external excitation, e.g. via a photon ← photon detector
2. thermal excitation
3. impurities

The Fermi Energy

The Fermi energy E_F determines the concentration of thermally excited electrons in the conduction band.



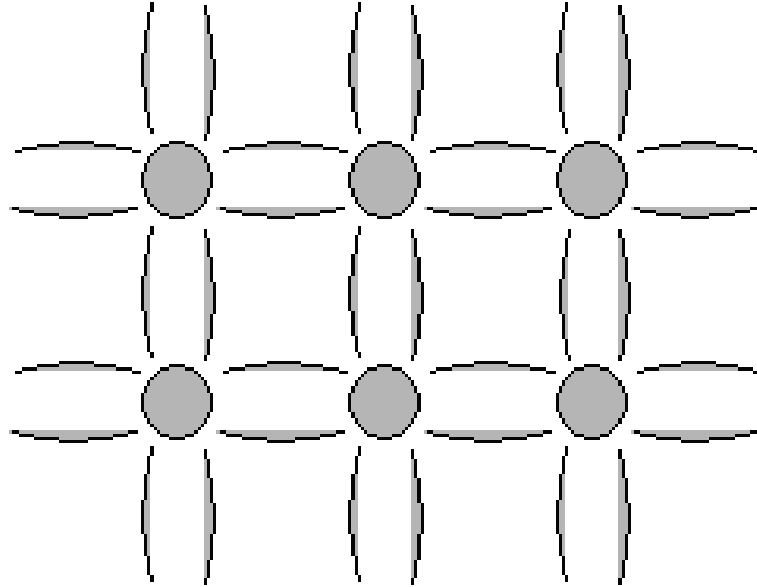
$$n_0 = N_c f(E_c) \quad \text{where} \quad N_c = 2 \left(\frac{2\pi m_{eff} kT}{h^2} \right)^{3/2}$$

$$\text{and} \quad f(E_c) = \frac{1}{1 + e^{(E_c - E_F)/kT}} \quad \begin{matrix} E_c - E_F \gg kT \\ \approx e^{-(E_c - E_F)/kT} \end{matrix}$$

Intrinsic Photoconductors

The Basic Principle

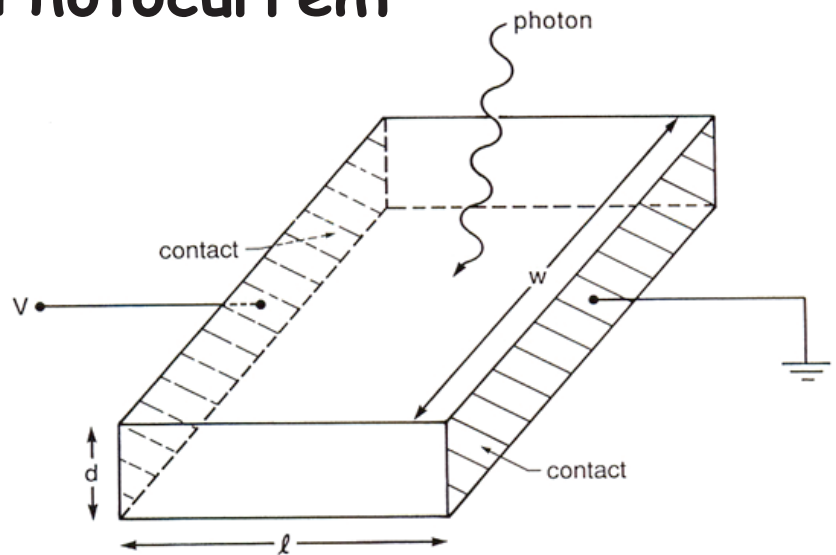
- E_v lifts e^- into conduction band
- electric field \bar{E} drives charges to electrodes
- few charge carriers \rightarrow high resistance



The Photocurrent

Conductivity:

$$\sigma = \frac{1}{R_d} \frac{l}{wd} = qn_0\mu_n$$



where:

R_d = resistance

w, d, l = geometric dimensions

q = electric charge

n_0 = number of charge carriers $n = \frac{\phi\eta\tau}{wdl}$

ϕ = photon flux

τ = mean lifetime before recombination

μ_n = electron mobility \sim mean time between collisions.

Important Quantities and Definitions

$$\text{Quantum efficiency } \eta \equiv \frac{\# \text{ absorbed photons}}{\# \text{ incoming photons}}$$

$$\text{Responsivity } S \equiv \frac{\text{electrical output signal}}{\text{input photon power}}$$

$$\text{Wavelength cutoff: } \lambda_c = \frac{hc}{E_g} = \frac{1.24 \mu\text{m}}{E_g [\text{eV}]}$$

$$\text{Photo-current: } I_{ph} = q\phi\eta G$$

$$\text{Photoconductive gain } G: G = \frac{I_{ph}}{q\phi\eta} = \frac{\tau}{\tau_t} = \frac{\text{carrier lifetime}}{\text{transit time}}$$

The *product ηG* describes the probability that an incoming photon will produce an electric charge that will penetrate to an electrode.

Limitations of Intrinsic Semiconductors

- short wavelength cutoffs $\lambda_c = \frac{hc}{E_g}$
 - Germanium: 1.85 μm
 - Silicon: 1.12 μm
 - GaAs: 0.87 μm
- non-uniformity of material
- problems to make good electrical contacts to pure Si
- difficult to "keep clean" and minimize Johnson noise

Extrinsic Photoconductors

Extrinsic Semiconductors

Solution: add impurities at low concentration to provide excess electrons → much reduced bandgap → longer wavelength cutoff

n-doped silicon

p-doped silicon

Impurity	Type	Ge	Si
		Cutoff wavelength λ_c (μm)	Cutoff wavelength λ_c (μm)
Al	p		18.5 ^a
B	p	119 ^b	28 ^a
Be	p	52 ^b	8.3 ^a
Ga	p	115 ^b	17.2 ^a
In	p	111 ^b	7.9 ^a
As	n	98 ^b	23 ^a
Cu	p	31 ^b	5.2 ^a
P	n	103 ^b	27 ^a
Sb	n	129 ^b	29 ^a

Example: addition of boron to silicon in the ratio 1:100,000 increases its conductivity by a factor of 1000!

Problems: absorption coefficients much less than for intrinsic photoconductors → low QE → active volumes (pixels) must be large

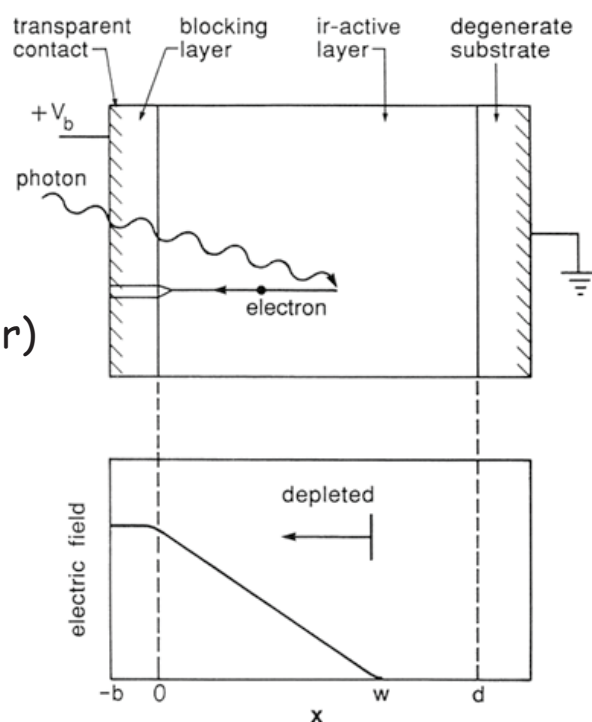
Blocked Impurity Band (BIB) Detectors

Solution: use separate layers to optimize the optical and electrical properties independently:

IR-active layer: heavily doped

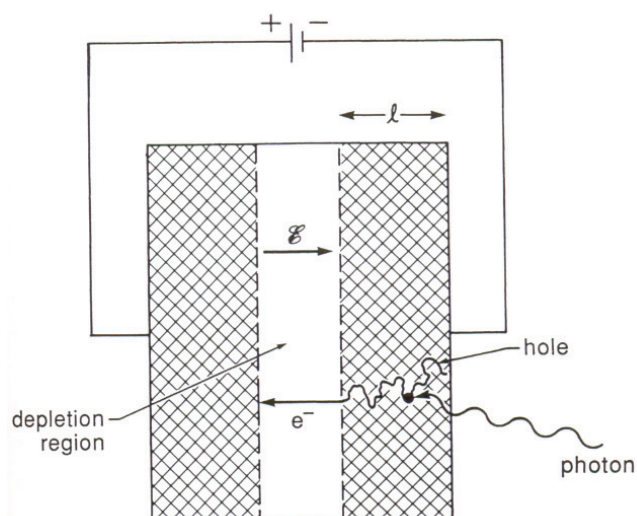
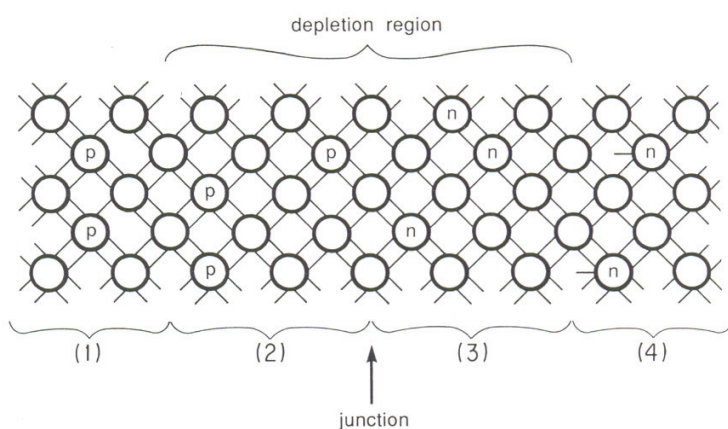
Blocking layer: thin layer of high purity
(intrinsic photoconductor)

Typical species are *Si:As* or *Si:Sb BIBs*



Photodiodes

- Based on junction between *two* oppositely doped zones
- The two adjacent zones create a depletion region with high impedance



1. Photon gets absorbed e.g. in the p-type part
2. Absorption creates an e^- -hole pair
3. The e^- diffuses through the material
4. Voltage drives the e^- across the depletion region \rightarrow photo-current

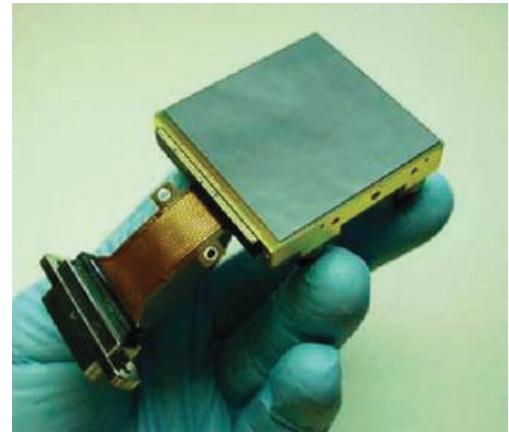
Example: The Teledyne HAWAII-2RG

Parameter	Specification
Detector technology	HgCdTe or Si PIN
Detector input circuit	SFD
Readout mode	Ripple
Pixel readout rate	100 kHz to 5MHz (continuously adjustable)
Total pixels	2048 x 2048
Pixel pitch	18 μm
Fill factor	$\geq 98\%$
Output ports	Signal: 1, 4, 32 selectable guide window and reference
Spectral range	0.3 - 5.3 μm
Operating temperature	$\geq 30\text{K}$
Quantum efficiency (array mean)	$\geq 65\%$
Charge storage capacity	$\geq 100,000e^-$
Pixel operability	$\geq 95\%$
Dark current (array mean)	$\leq 0.1 e^-/\text{sec}$ (77K, 2.5 μm)
Read noise (array mean)	$\leq 15 e^-$ CDS @ 100 kHz
Power dissipation	$\leq 4 \text{ mW}$ @ 100 kHz

See

<http://www.rsc.rockwell.com/imaging/hawaii2rg.html> for more info

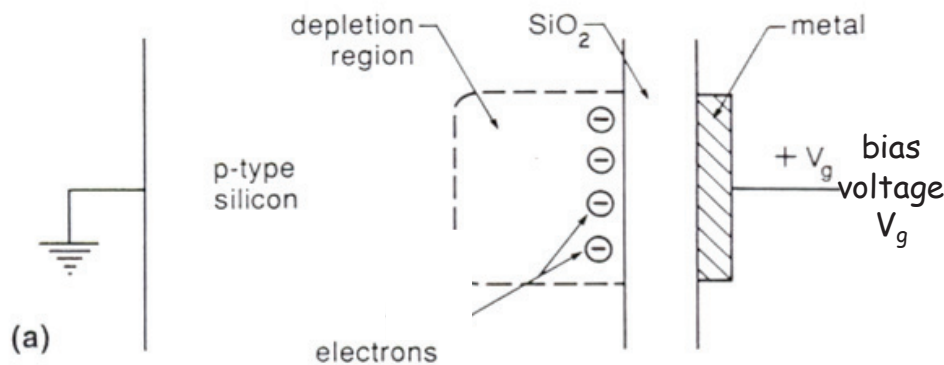
Can also be combined to a 2x2 mosaic



Charge Coupled Devices (CCDs)

CCDs = array of integrating capacitors.

Pixel structure: metal "gate" evaporated onto SiO_2 (isolator) on silicon = MOS



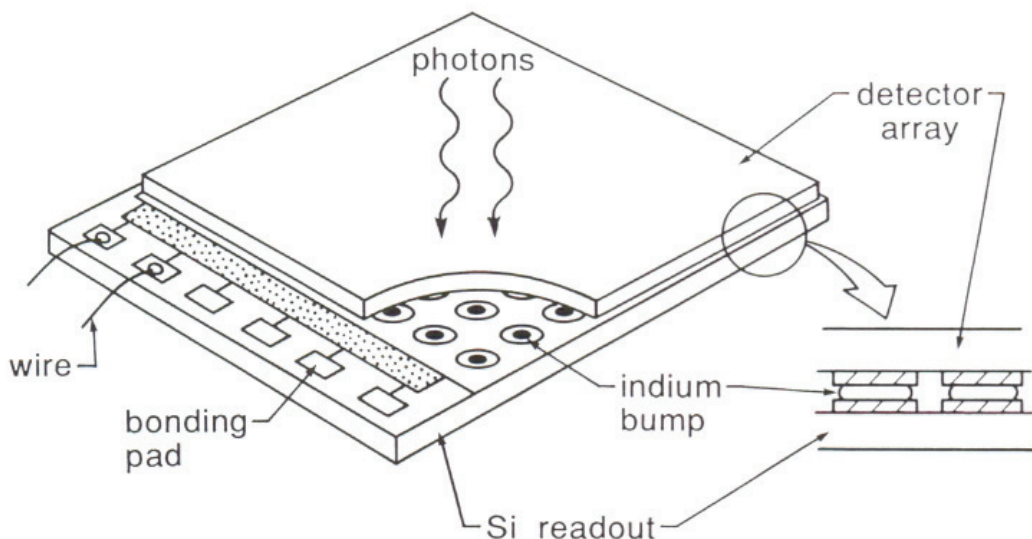
1. photons create free e^- in the photoconductor
2. e^- drift toward the electrode but cannot penetrate the SiO_2 layer
3. e^- accumulate at the Si-SiO₂ interface
4. the total charge collected at the interface is a measure of the number of photons during the exposure
5. \rightarrow read out the number of e^-

Readout and Operation

I. Infrared Arrays

Infrared Arrays - Construction

1. Produce a grid of readout amplifiers
2. Produce a (matching mirror image) of detector pixels
3. Deposit **Indium bumps** on both sides
4. Squeeze the two planes together → **hybrid arrays**
5. The Indium will flow and provide electrical contact



Multiplexers

Multiplexing: "Pixel signals \rightarrow Sequential output lines"

MUX Tasks:

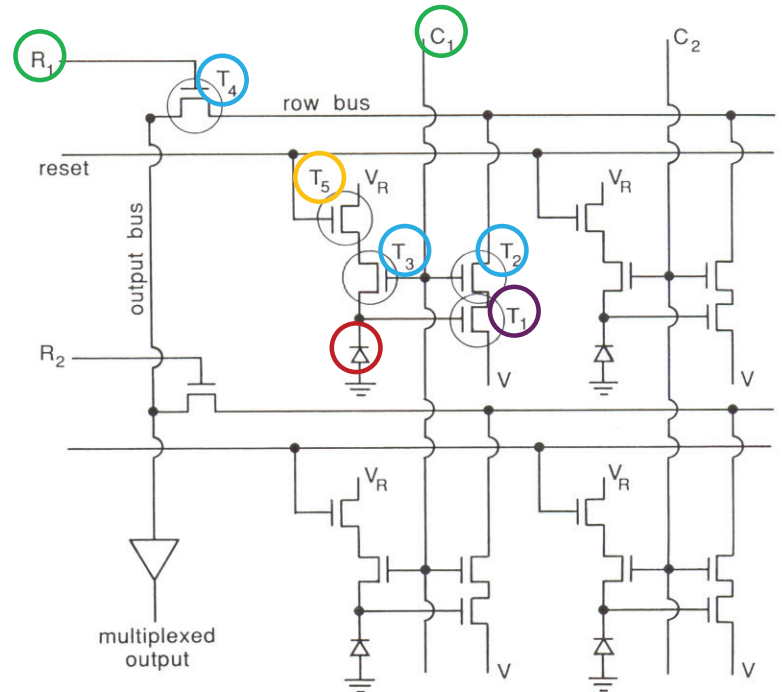
- address a column of pixels by turning on their amplifiers
- pixels in other columns with power off will not contribute a signal

Signal at photodiode \rightarrow gate T_1

Readout uses row driver R_1 and column driver C_1 to close the switching transistors T_2, T_3, T_4 .

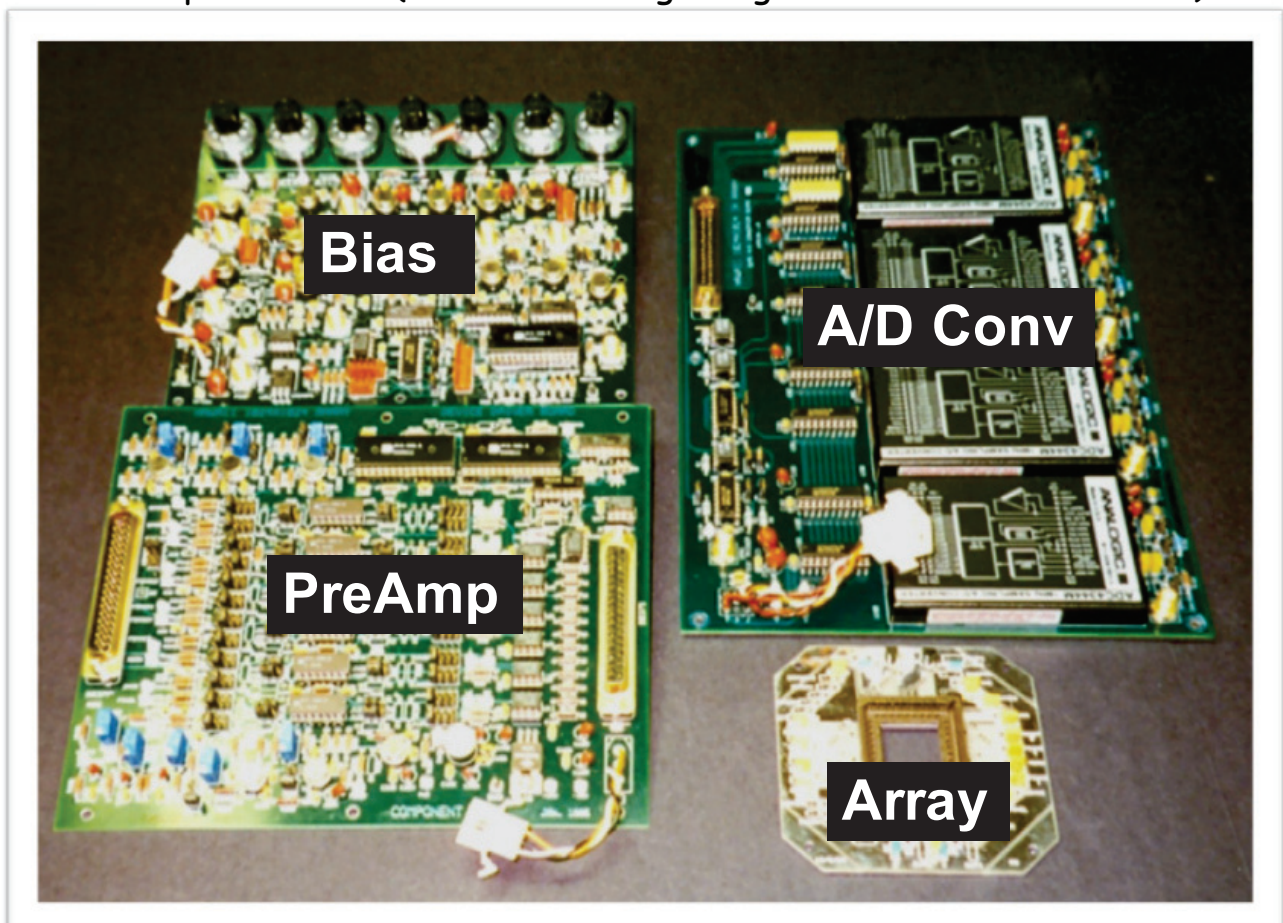
\rightarrow Power to $T_1 \rightarrow$ signal to the output bus

Reset: connect V_R via T_5 and T_3 .



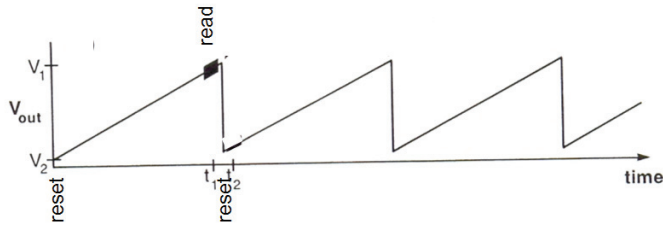
Elements of a Detector Electronics System

Example: PHARO (the Palomar High Angular Resolution Observer)



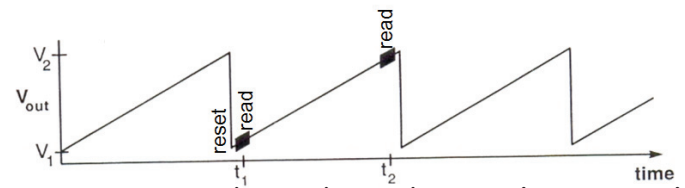
IR Array Read Out Modes

Single Sampling



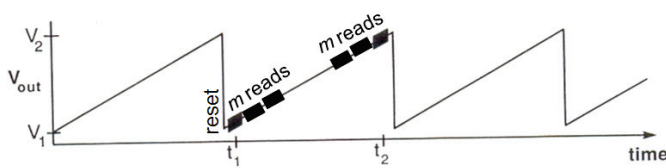
- most simple approach
- does not remove kTC noise
- measures the absolute signal level

Reset-Read-Read



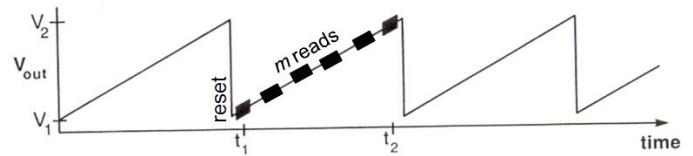
- Resets, reads and reads pixel-by-pixel
- Signal = Read(2) - Read(1)
- best correlation, no reset noise
- but requires frame storage
- reduced dynamical range (saturation!)

(Multiple) Fowler Sampling



- similar to reset-read-read ...
- ... but each read is repeated m times
- Signal = mean(read2) - mean(read1)
- Reduces readout noise by \sqrt{m} over RRR

Sample-up-the-ramp Fitting



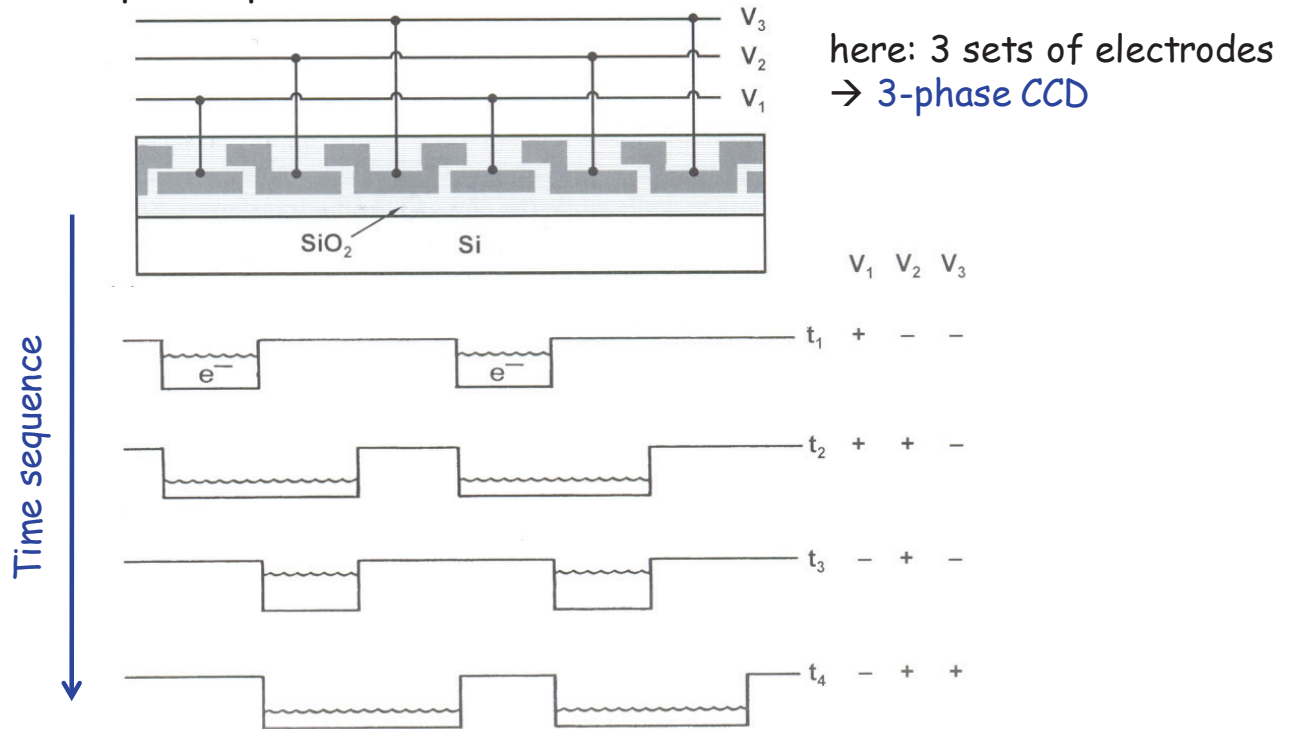
- m equidistant reads during integration
- linear fit \rightarrow "slope"
- reduces readout noise by \sqrt{m}
- particularly useful in space (cosmics!)

Readout and Operation

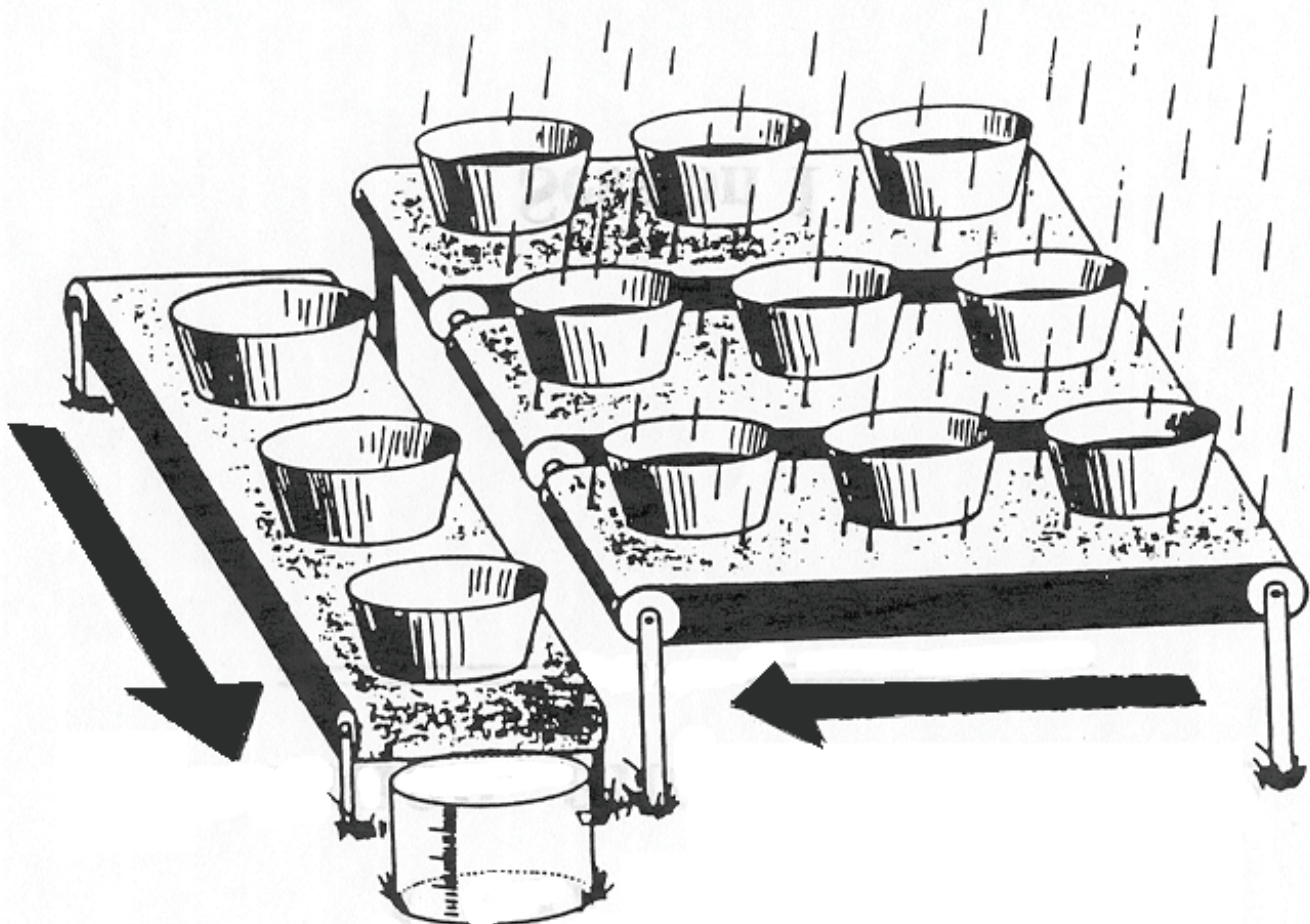
II. CCDs

Charge Coupled Readouts

Collected charges are passed along the columns to the edge of the array to the output amplifier.



Be aware of charge transfer (in-)efficiencies (CTEs) due to electrostatic repulsion, thermal diffusion and fringing fields.



Charge Transfer Efficiency (CTE)

Time-dependent mechanisms that influence the CTE:

1. **Electrostatic repulsion** causes electrons to drift to the neighbouring electrode with time constant for charge transfer τ_{SI} .
2. **Thermal diffusion** drives electrons across the storage well at τ_{th}
3. "Fringing fields" due to dependency of the well on the voltages of neighbouring electrodes (τ_{ff}).

Approximation for the CTE of a CCD with m phases: $CTE = \left(1 - e^{-t/\tau}\right)^m$

Noise from **charge transfer inefficiency**: $\varepsilon = (1-CTE)$

CCDs and IR Arrays are fundamentally different!

CCDs:

- destructive reads
- charges are physically shifted to the output line
- shutter determines exposure time

IR arrays:

- non-destructive reads
- readout requires sophisticated multiplexer circuit
- multiplexer readout addresses individual pixels directly
- read/reset determines exposure time

Orthogonal Transfer CCDs (OTCCD)

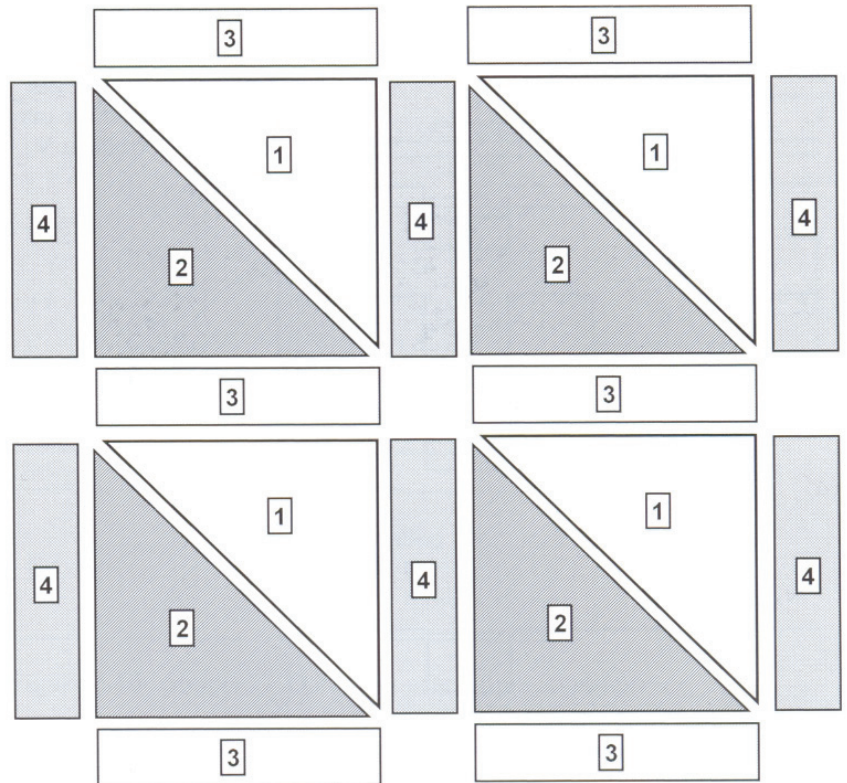
For TDI it would be desirable to *move the charges in any direction* to follow the image motion. This can be done with the **OTCCD**.

OTCCD operation:

To move a charge to the right, `3' is negative to act as channel stop, `1', `2', and `4' are operated as a conventional CCD.

To move a charge up, `4' is negative to act as channel stop, `1', `2', and `3' are operated as a conventional CCD.

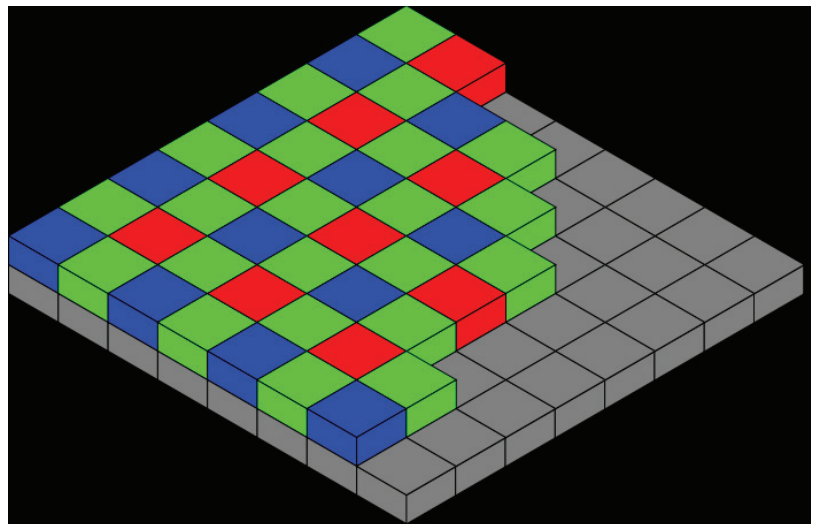
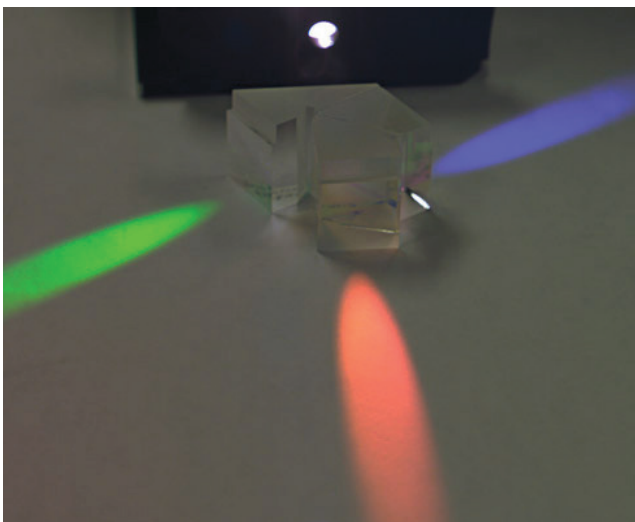
Moving to the opposite directions: reversing the clocking.



Side note: CCD Color Images

Essentially three ways to do it (from Wikipedia):

1. Take three exposures through three filters subsequently - only works for fixed targets (*standard for astronomy*).
2. Split the input beam in three channels, each with a separate and optimized CCD (*very expensive cameras*).
3. Use a Bayer mask over the CCD - each subset of 4 pixels has one filtered red, one blue, and two green.



Detector Noise

The main Noise Components

G-R noise $\langle I_{G-R}^2 \rangle = 4q^2 \phi \eta G^2 \Delta f$

fundamental **statistical noise** due to the **Poisson statistics** of the photon arrival → transferred into the statistics of the **generated** and **recombined** holes and electrons.

Johnson or kTC noise $\langle I_J^2 \rangle = \frac{4kT}{R} \Delta f$

fundamental **thermodynamic noise** due to the thermal motion of the charge carriers. Consider a photo-conductor as an RC circuit. Since $\langle Q^2 \rangle = kTC$, the charge noise is also called kTC noise or reset noise.

1/f noise $\langle I_{1/f}^2 \rangle \propto \frac{I^2}{f} \Delta f$

increased **noise at low frequencies**, due to bad electrical contacts, temperature fluctuations, surface effects (damage), crystal defects, and JFETs, ...

The total noise in the system is: $\langle I_N^2 \rangle = \langle I_{G-R}^2 \rangle + \langle I_J^2 \rangle + \langle I_{1/f}^2 \rangle$

BLIP and NEP

Operationally, background-limited performance (BLIP)

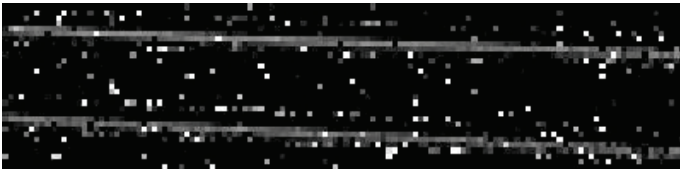
is always preferred: $\langle I_{G-R}^2 \rangle \gg \langle I_J^2 \rangle + \langle I_{1/f}^2 \rangle$

The noise equivalent power (NEP) is the signal power that yields an RMS S/N of unity in a system of $\Delta f = 1$ Hz:

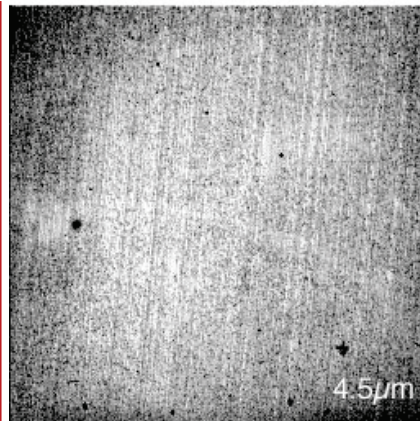
$$NEP_{G-R} = \frac{2hc}{\lambda} \left(\frac{\varphi}{\eta} \right)^{1/2}$$

In BLIP the NEP can only be improved by increasing the quantum efficiency η .

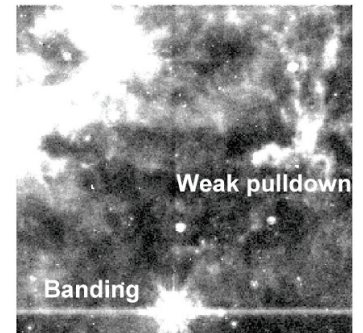
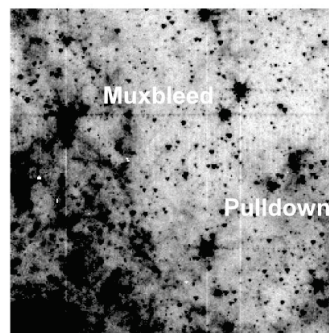
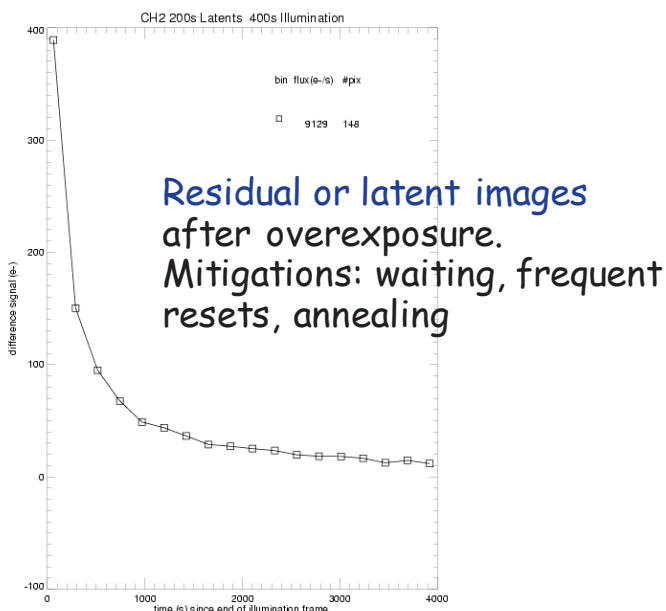
Detector Artefacts (1)



Dead, hot and rogue pixels.
Mitigation: subtract off-source image and/or reduce bias voltage



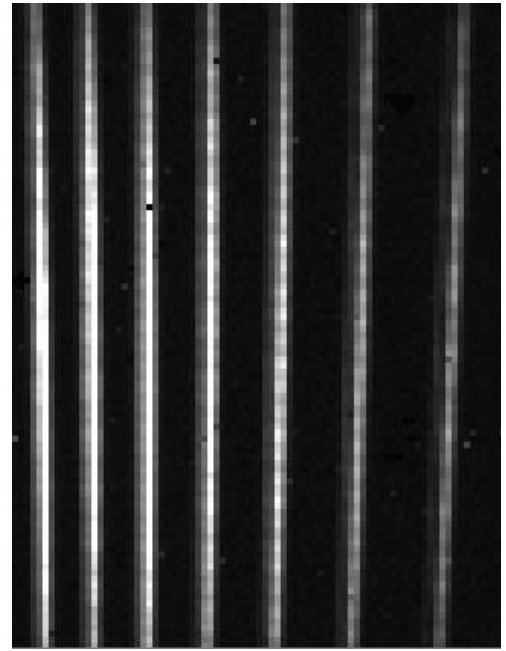
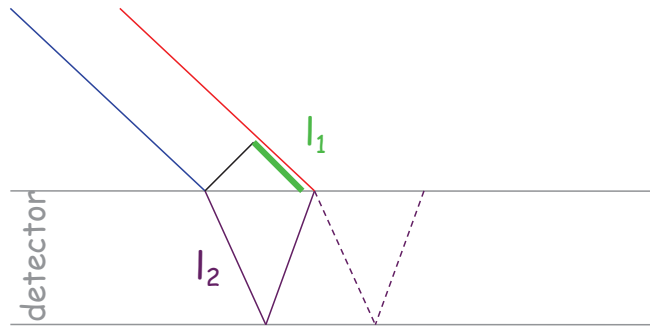
Fixed pattern noise.
Mitigation: "flat-fielding"



Muxbleed, pulldown and banding.
Mitigation: avoid bright sources, short exposures.

Detector Artefacts (2): Fringing

In spectrographs: photons reflect off the back of the detector and interfere with the incoming light.



If the phase difference between l_1 and $n \cdot l_2$ is an even multiple of π constructive interference occurs. If an odd multiple destructive interference occurs \rightarrow fringes = wave pattern.

Flatfielding Techniques

General Flatfielding

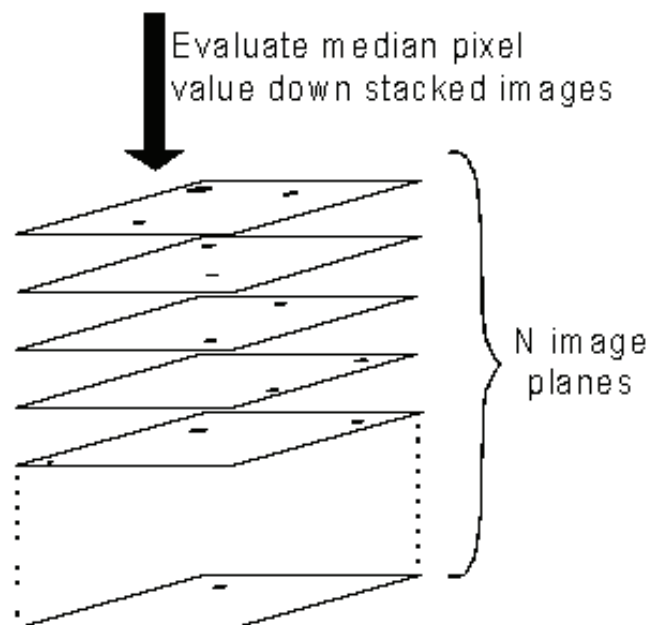
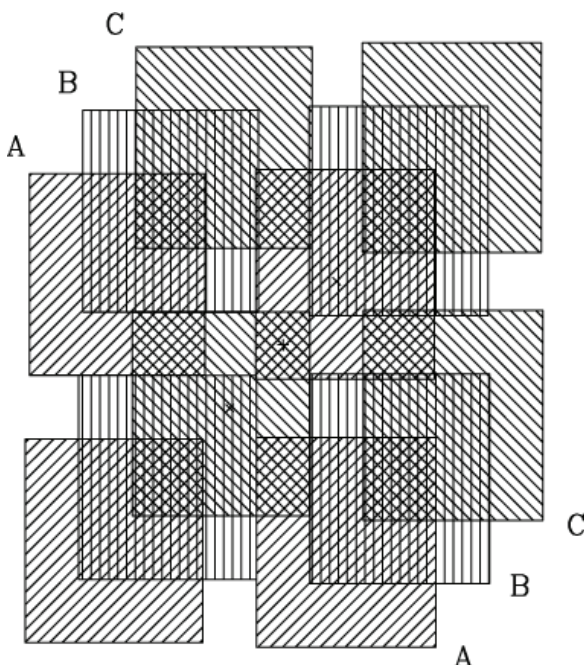
Detector response (QE, bias) varies slightly from pixel to pixel → image has "structure", even with flat illumination
→ flat-fielding; common methods are:

1. **Dome flats:** illuminate a white screen within the dome (can be done during the day, but may introduce spectral artifacts)
2. **Twilight flats:** observe the twilight sky at two times during sunrise or sunset (high S/N but time is often too short to get FFs for all filters)
3. **Sky flats:** use the observations themselves (spectrally best, but often low S/N)

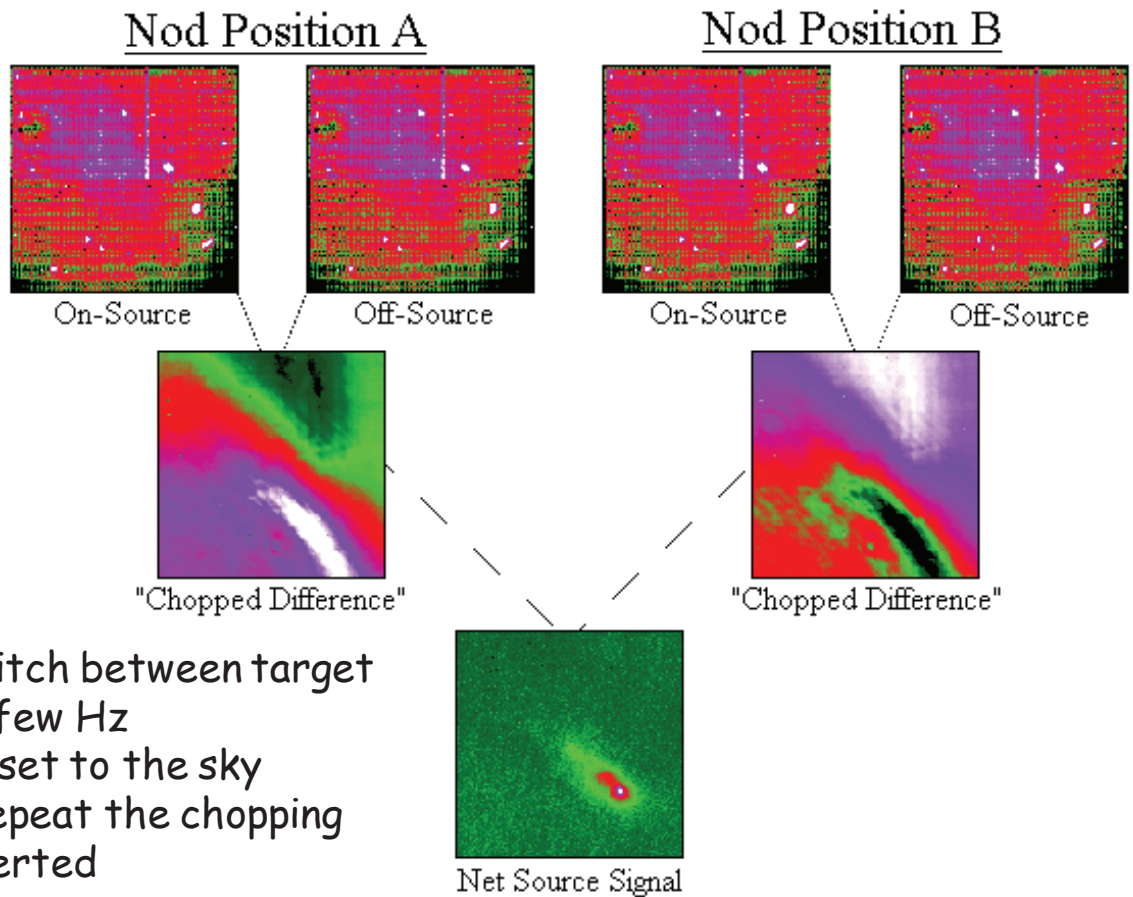
In all cases: use the difference between two flux levels F_1, F_2 to compute the flatfield $FF = \left(\frac{F_1 - F_2}{\text{median}(F_1 - F_2)} \right)^{-1}$ with which all images have to be multiplied.

Dithering / Jittering

1. Observe the same field with many exposures, each offset by a small amount
2. Combine the image e.g., via median filtering



Chopping / Nodding



Part I

Photon Detectors

Part II

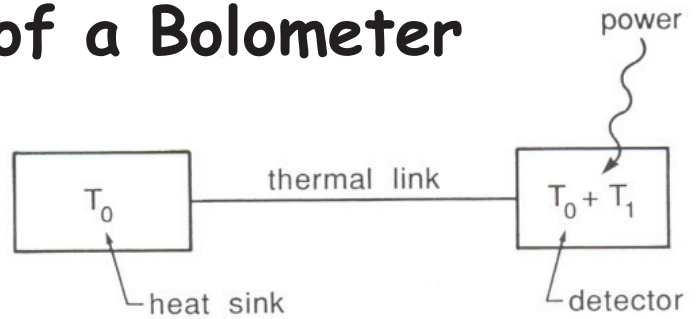
Thermal Detectors

Part III

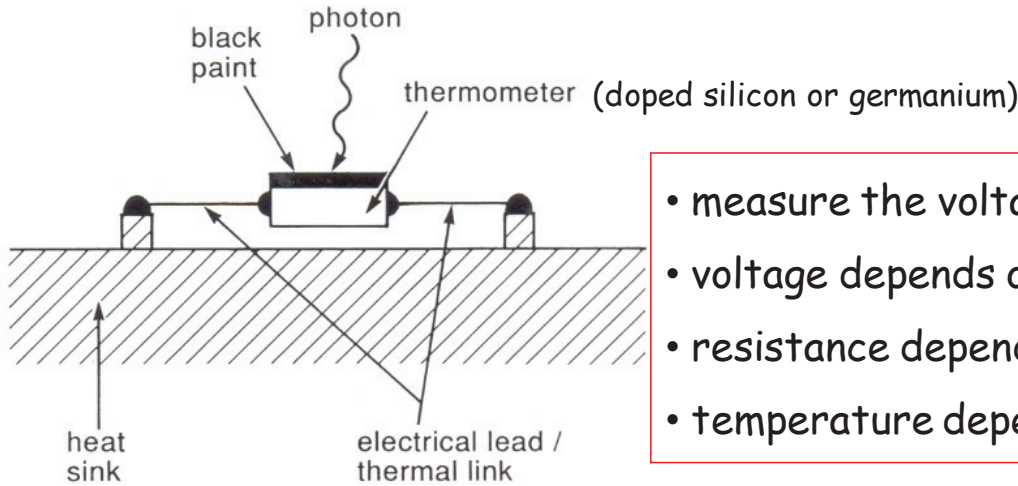
Coherent Receivers

Basic Principle of a Bolometer

A detector with thermal heat capacity C is connected via a **thermal link** of thermal conductance G to a heat sink of temperature T_0 .



The **total power** absorbed by the detector is:
$$P_T(t) = GT_1 + C \frac{dT_1}{dt}$$



- measure the voltage across thermo.
- voltage depends on resistance
- resistance depends on temperature
- temperature depends on photon flux

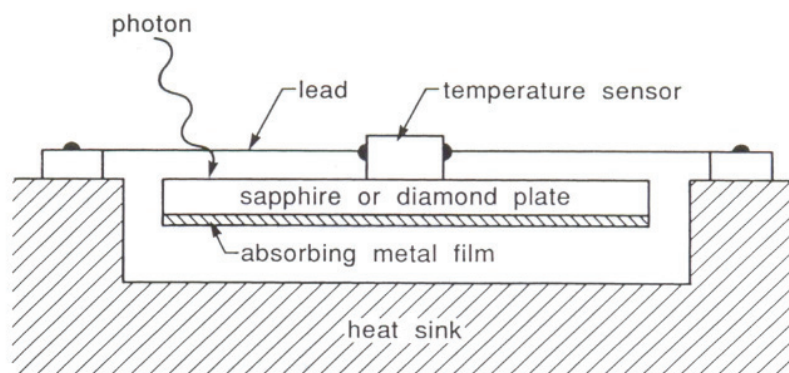
Bolometers are especially for the far-IR/sub-mm wavelength range!

QE and Composite Bolometers

In some cases Si bolometers with high impurity concentrations can be very efficient absorbers.

In many cases, however, the QE is too low. Solution: enhance absorption with black paint - but this will increase the heat capacity.

A high QE bolometer for far-IR and sub-mm would have too much heat capacity → **composite bolometers**.

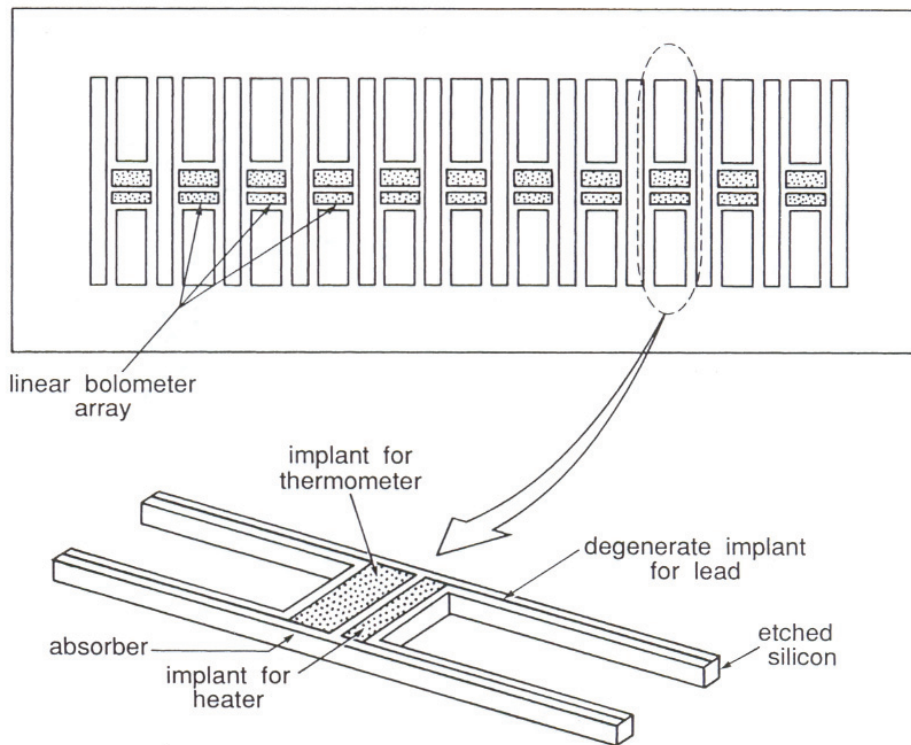


The heat capacity of the blackened sapphire plate is only 2% of that of Ge.

Etched Bolometers

The bolometer design has been revolutionized by precision etching techniques in Si

Thermal time response $\sim C/G \rightarrow$ small structures minimize the heat capacity C by reducing the volume of material.

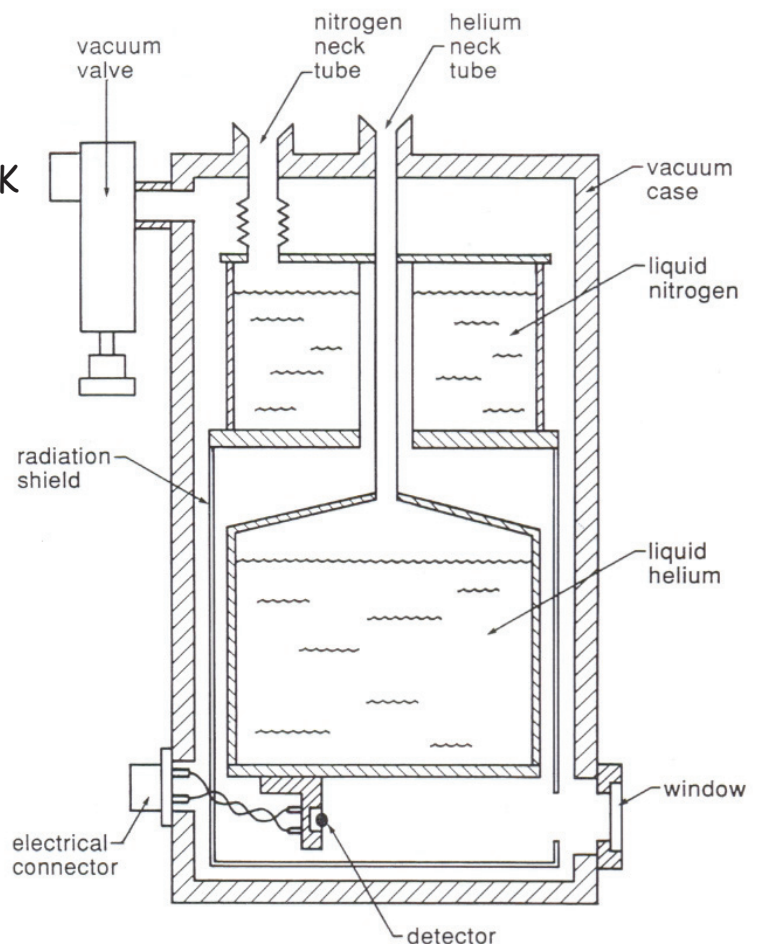


Low Operating Temperatures

Four standard options to cool:

1. ^4He dewar (air pressure) $\rightarrow T=4.2\text{K}$
2. ^4He dewar (pumped) $\rightarrow 1\text{K} < T < 2\text{K}$
3. ^3He (closed-cycle) refrigerator $\rightarrow T \sim 0.3\text{K}$
4. adiabatic demagnetization refrigerator $\rightarrow T \sim 0.1\text{K}$

Simplest solution is to use a two-stage helium dewar (here: model from Infrared Laboratories, Inc.)

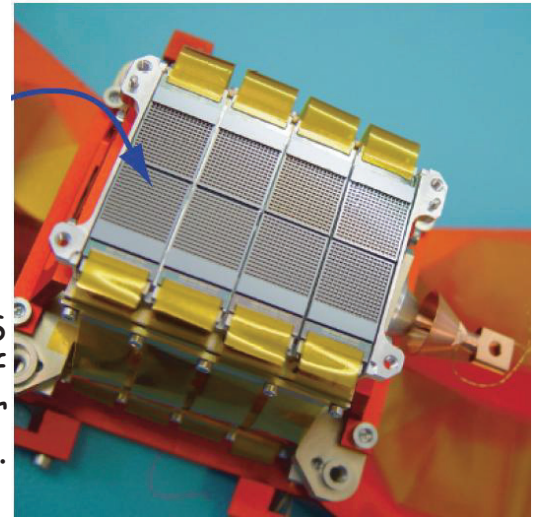


Bolometers - an Overview



The "single pixel" Ge:Ga bolometer invented in 1961 by Frank Low

Herschel / PACS bolometer: a cut-out of the 64x32 pixel bolometer array assembly.

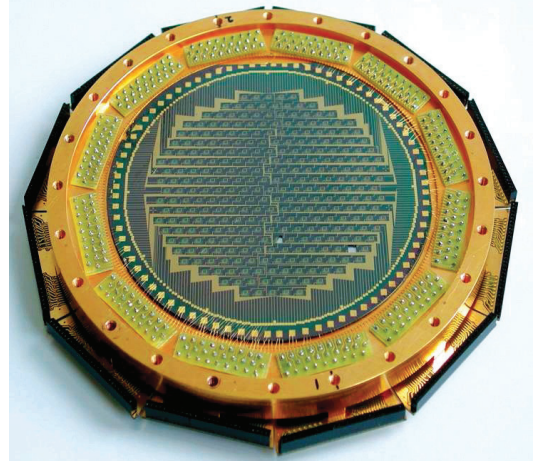


LABOCA - the multi-channel bolometer array for APEX operating in the 870 μm (345 GHz) atmospheric window.

The signal photons are absorbed by a thin metal film cooled to about **280 mK**.

The array consists of 295 channels in 9 concentric hexagons.

The array is under-sampled, thus special mapping techniques must be used.



Part I

Photon Detectors

Part II

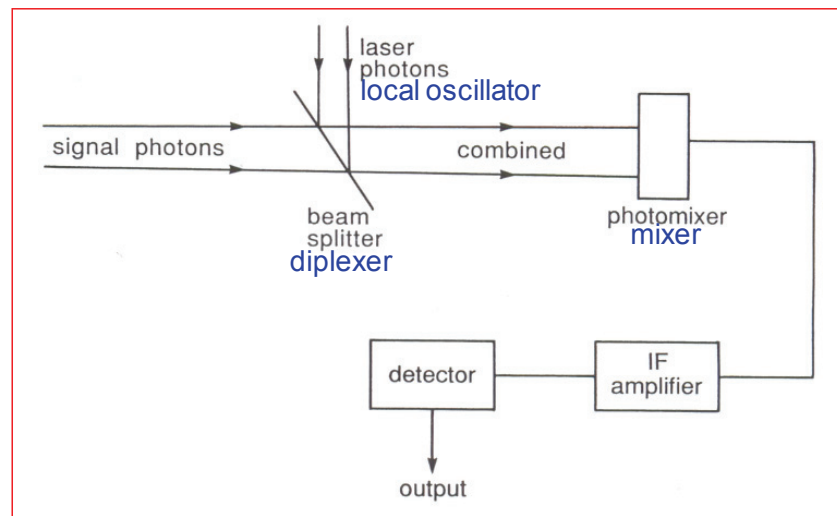
Thermal Detectors

Part III

Coherent Receivers

Basic Principle

Already introduced previously → [see Radio Techniques lecture](#)



Performance Characteristics

Performance Comparison Bolometer ↔ Heterodyne Receiver

Case 1: Bolometer operating at BLIP and heterodyne receiver operating in the thermal limit ($h\nu \ll kT$)

→ the bolometer will perform better

This is always true, except for measurements at high spectral resolution, much higher than the IF bandwidth.

Case 2: detector noise-limited bolometer and a heterodyne receiver operating at the quantum limit ($h\nu \gg kT$).

→ the heterodyne receiver will outperform the bolometer.

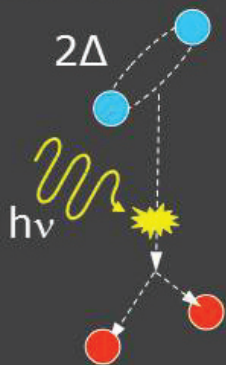
In the case of narrow bandwidth and high spectral resolution the heterodyne system will always win.

Outlook: The next Generation of Detectors

(courtesy: Jochem Baselmans/SRON)

MKIDS - Physical Principle

Superconductor has 2 types of charge carriers



Cooper Pairs

Supercurrent

Inductance
 $i\omega L(P_{\text{sky}})$

Quasiparticles

Normal current

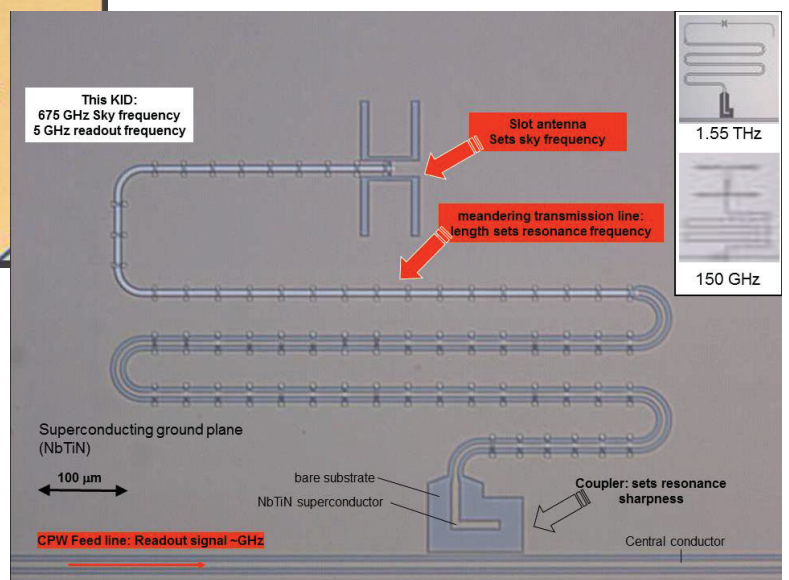
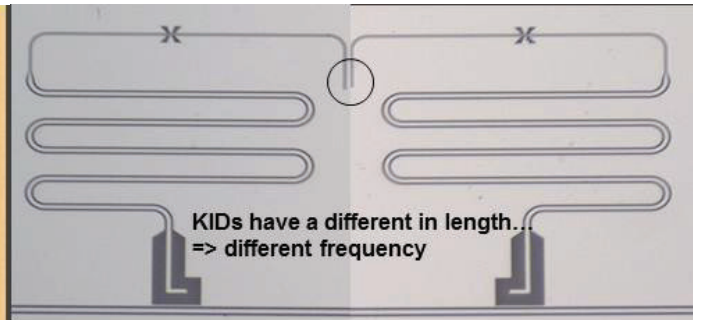
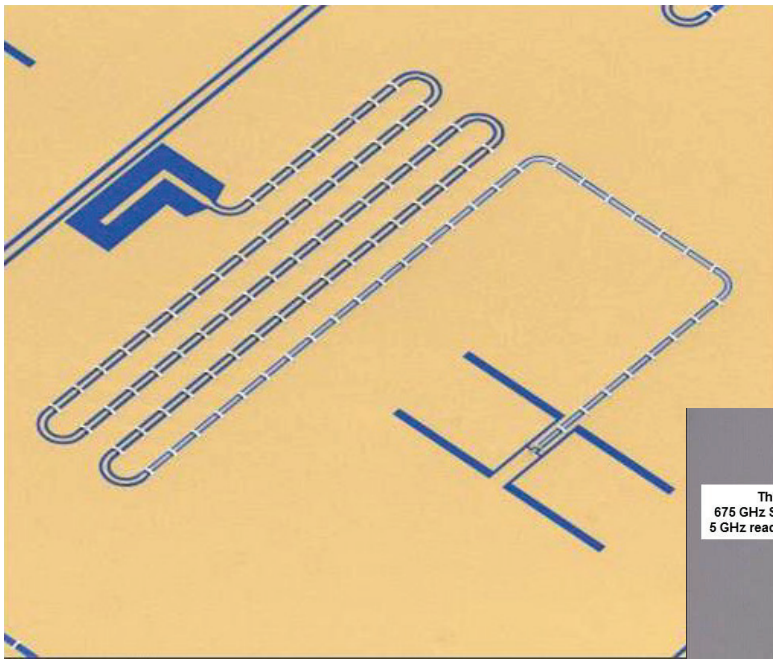
Resistance
 $R(P_{\text{sky}})$

KID

Resonator measures changes in complex surface impedance due to sky signal

KID = Kinetic Inductance Detector

MKIDS - Construction



MKIDS - Operating Principle

